1-ARYL-3-NITROPROP-2-EN-1-ONES IN REACTIONS WITH SUBSTITUTED HYDRAZINES

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1-Aryl-3-nitroprop-2-en-1-ones, containing in their structure a C=C bond conjugated with two acceptor substituents, are attractive substrates in reactions with nucleophilic reagents.

It has been previously shown that structurally similar alkyl 3-nitroacrylates react with carboxylic acid hydrazides and hydrazinecarboxamide to form Michael adducts [1, 2].

We have studied the interaction of 1-aryl-3-nitroprop-2-en-1-ones **1a**, **b** [3] with representatives of substituted hydrazines – benzohydrazide **2a** and *N*-phenylhydrazinecarboxamide **2b**. The reaction proceeds in ethanol solution at room temperature and leads to the formation of corresponding aza-Michael adducts **3a-d** with yields up to 92%.

R¹
$$\xrightarrow{O}$$
 \xrightarrow{H} \xrightarrow{H} \xrightarrow{P} \xrightarrow{EtOH} \xrightarrow{R} \xrightarrow{EtOH} \xrightarrow{NH} \xrightarrow{NH} \xrightarrow{NH} \xrightarrow{NH} $\xrightarrow{Sa-d}$ $\xrightarrow{R^1 = H \ (1a, 3a, b), Me \ (1b, 3c, d); R^2 = Ph \ (2a, 3a, c), NHPh \ (2b, 3b, d)}$

The structure of synthesized compounds **3a-d** was characterized by a complex of physicochemical methods (IR, NMR ¹H, ¹³C spectroscopy with ¹H–¹³C HMQC and ¹H–¹³C HMBC experiments). The structural feature of adducts **3a-d** is the presence of an asymmetric carbon atom, which causes the manifestation diastereotopy effect in the ¹H NMR spectra of protons of the CH₂NO₂ group.

This work was financially support by the Ministry of Education of Russia (project no. VRFY-2023-0003).

References:

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